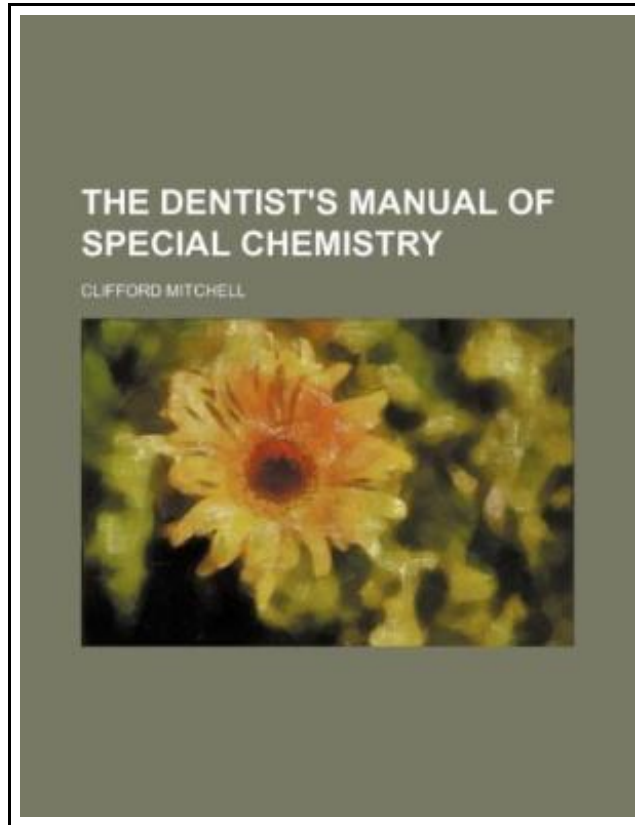


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1887 Excerpt: .but the actual molecular formula contains twice the number of atoms, or $C_2H_4O_3$ Besides empirical and molecular formulae, others called rational, constitutional, structural or graphic are used. The molecular formula of acetic acid is $C_2H_4O_2$, but the formula $HC_2H_3O_2$ shows that acetic acid, like nitric acid HN_3 , is monobasic, containing one atom of hydrogen, which can be replaced by an atom of a metal; hence $HC_2H_3O_2$ is called a constitutional formula. 7. Radicals or residues. These are expressions for unsaturated groups of atoms known to enter as a whole into different compounds, but having no separate existence. Water H_2O is a saturated compound that is the one atom of oxygen--which is a dyad, and may be said therefore to have two points of attraction--combines with two of hydrogen; and therefore has both its points of attraction satisfied. If now one atom of H be taken from H_2O , there is left the group of atoms HO , which is called a radical, as it consists of an atom of oxygen, in which but one point of attraction is actually saturated, the second one not being provided for; moreover, this group HO occurs in many compounds--as, for example, in the hydrates, as potassium hydrate, KHO , etc. The equivalence of radicals depends upon the number of points of attraction unprovided for: carbon requires four atoms of hydrogen to provide for its points of attraction; therefore, CH_3 would be a monad radical, CH_2 a dyad, CH a triad. 8. Chains: the...



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